

***N,N*-Dibenzoyl-4-chloroaniline**

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Key indicators

Single-crystal X-ray study

T = 183 K

Mean $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$

R factor = 0.054

wR factor = 0.131

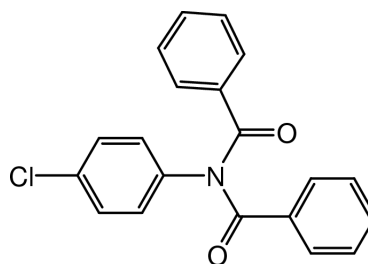
Data-to-parameter ratio = 13.4

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound, $\text{C}_{20}\text{H}_{14}\text{ClNO}_2$, the three aromatic rings are almost orthogonal to one another. The molecular packing is stabilized by weak $\text{C}-\text{H}\cdots\pi$ interactions and van der Waals forces.

Comment

Photoinduced electron transfer (PET) reactions of organic compounds have been receiving extensive investigation in recent years (Scheinbaum, 1964; Kavarnos & Turro, 1986; D'Auria *et al.*, 1996). However, PET reactions of nitroaromatic compounds, as electron acceptors, with different types of donors have not been much investigated. As a part of our ongoing research interest in PET reactions of various electron donor–acceptor combinations (Xue *et al.*, 2000, 2001; Zhang *et al.*, 2000), we have recently investigated the photo-induced reactions between nitroaromatic compounds and electron-rich alkenes, such as 1,2-diphenylacetylene. The title compound, (I), a dibenzoylaniline derivative, was isolated from the photoreaction of 4-chloronitrobenzene with 1,2-diphenylacetylene.



(I)

The bond lengths and angles in (I) have normal values (Allen *et al.*, 1987). The three aromatic rings *A* (C1–C6), *B* (C8–C13) and *C* (C15–C20) are almost orthogonal to one another (Fig. 1). The dihedral angles between *A* and *B*, *A* and *C*, *B* and *C* are 74.7 (1), 89.8 (1) and 81.6 (1)°, respectively. The O1/C7/C8/N1 amide plane is twisted by an angle of 30.2 (1)° from the benzoyl phenyl ring *B*, whereas the other amide plane, O2/C14/C15/N1, is twisted by 19.9 (1)° from ring *C*. The molecular packing in the crystal is stabilized by weak $\text{C}-\text{H}\cdots\pi$ interactions involving aromatic rings *B* and *C* (Table 2). In Table 2, $\text{Cg}(B)$ and $\text{Cg}(C)$ denote the centroids of rings *B* and *C*, respectively.

Experimental

The title compound was prepared by the photo-induced reaction of an acetonitrile solution of 4-chloronitrobenzene with an excess

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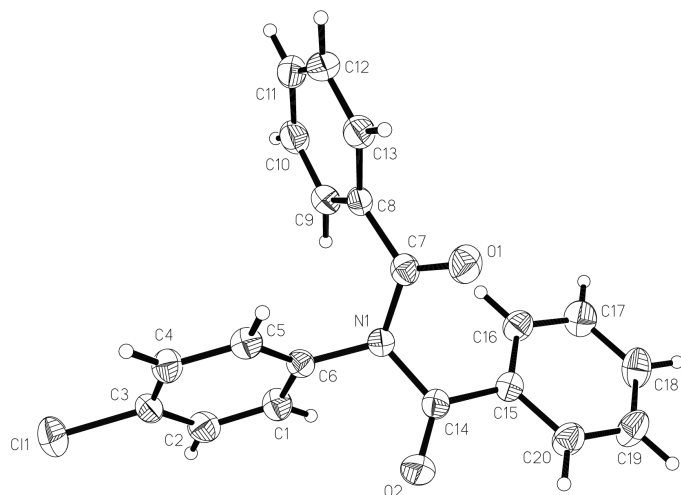


Figure 1
The structure of the compound (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.

amount of 1,2-diphenylacetylene and was isolated by column chromatography on silica gel. Single crystals of the compound were obtained by slow evaporation from a petroleum ether–ethyl acetate (2:1) solvent system.

Crystal data

$C_{20}H_{14}ClNO_2$
 $M_r = 335.77$
Monoclinic, $P2_1/n$
 $a = 10.2697$ (2) Å
 $b = 9.0419$ (2) Å
 $c = 17.9903$ (4) Å
 $\beta = 92.877$ (1)°
 $V = 1668.4$ (1) Å³
 $Z = 4$

$D_x = 1.337$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 3800 reflections
 $\theta = 2.2$ – 29.6 °
 $\mu = 0.24$ mm⁻¹
 $T = 183$ (2) K
Block, colorless
 $0.46 \times 0.20 \times 0.12$ mm

Data collection

Siemens SMART CCD area-detector diffractometer
 ω scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.898$, $T_{\max} = 0.972$
8993 measured reflections

2911 independent reflections
1684 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.087$
 $\theta_{\text{max}} = 25.0$ °
 $h = -12 \rightarrow 12$
 $k = -10 \rightarrow 10$
 $l = -21 \rightarrow 17$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.054$
 $wR(F^2) = 0.131$
 $S = 0.89$
2911 reflections
218 parameters
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0499P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.32$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.28$ e Å⁻³
Extinction correction: SHELXTL
Extinction coefficient: 0.036 (3)

Table 1

Selected geometric parameters (Å, °).

O1–C7	1.221 (3)	N1–C6	1.445 (3)
O2–C14	1.207 (3)	C7–C8	1.483 (4)
N1–C7	1.413 (3)	C14–C15	1.490 (4)
N1–C14	1.430 (3)		

Table 2

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C1–H1 \cdots Cg(B^i)	0.93	2.72	3.597 (2)	158
C10–H10 \cdots Cg(C^i)	0.93	2.85	3.489 (2)	127

Symmetry code: (i) $\frac{3}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z$.

The H atoms were fixed geometrically and were treated as riding on their parent C atoms, with $C-H = 0.93$ Å and $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 1990).

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References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- D'Auria, M., Esposito, V. & Mauriello, G. (1996). *Tetrahedron*, **52**, 14253–14272.
- Kavarnos, G. J. & Turro, N. J. (1986). *Chem. Rev.* **86**, 401–449.
- Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
- Scheinbaum, M. L. (1964). *J. Am. Chem. Soc.* **29**, 2200–2203.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (1990). *Acta Cryst.* **A46**, C-34.
- Xue, J., Zhang, Y., Wang, X.-L., Fun, H.-K. & Xu, J.-H. (2000). *Org. Lett.* pp. 2583–2586.
- Xue, J., Zhang, Y., Wu, T., Fun, H.-K. & Xu, J.-H. (2001). *J. Chem. Soc. Perkin Trans. 1*, pp. 183–191.
- Zhang, Y., Qian, S.-P., Fun, H.-K. & Xu, J.-H. (2000). *Tetrahedron Lett.* **41**, 8141–8145.